

Under Trifluoromethane, Below Equilibrium

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- 1. Boyle's Law: $P_1V_1 = P_2V_2$
- 2. Charles's Law: $V_1/T_1 = V_2/T_2$
- 3. Gay-Lussac's Law: $P_1/T_1 = P_2/T_2$
- 4. Avogadro's Law: $V/n = k$

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